

# STAFF3d-spin: Update on Atomic Layer Deposition for Magnetism & Spintronics

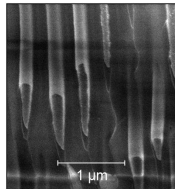
**Michal Staňo**

[michal.stano@ceitec.vutbr.cz](mailto:michal.stano@ceitec.vutbr.cz)



**CEITEC Magnetism seminar**

June 2, 2021



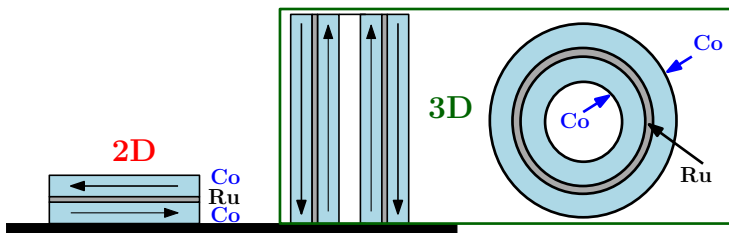
STAFF3d-spin project: <http://magnetism.ceitec.cz/staff3d-spin/>

# STAFF3d-spin project: multilayered nanotubes

Synthesis and investigation of  
**Synthetic Tubular AntiFerromagnets For 3D Spintronics**

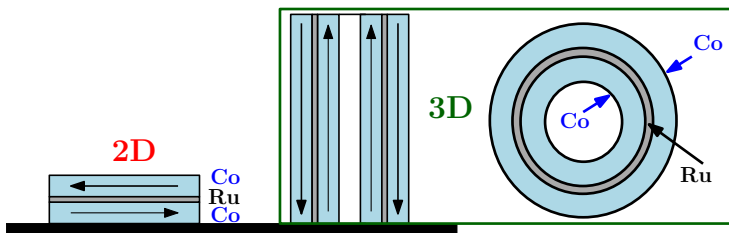
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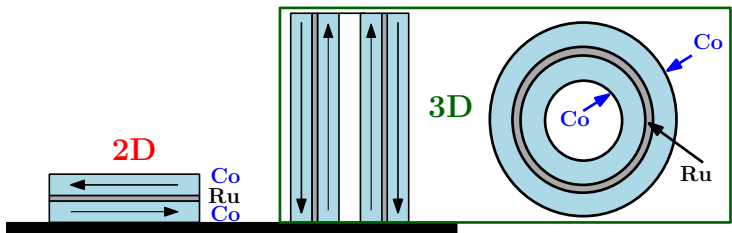


## Goals

- Preparation 3D vertical arrays of tubular SAF
- Test of interfaces suitability for spintronics (giant magnetoresistance)
- Investigation of individual magnetic nanotubes, tubular SAFs  
**experiments and simulations**

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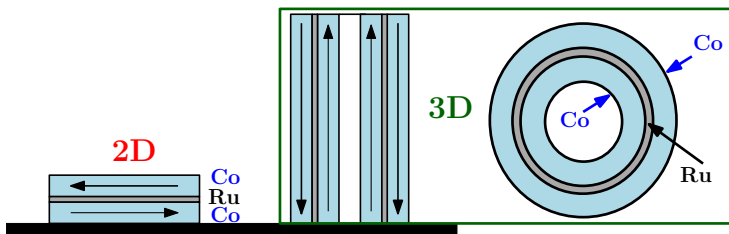


## Overview

- Previously: **ALD of planar Co** (magnetic, not fully optimized)

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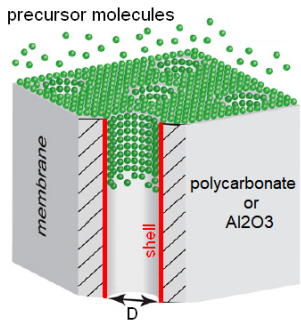
## Overview

- Previously: **ALD of planar Co** (magnetic, not fully optimized)
- Now: **ALD of planar Ru; planar Co on Ru**
- Now: **Nanotubes – optimization on HfOx**
- Next: Magnetic nanotubes, Magnetic/Ru/Magnetic layer, imaging, ...

# Outline of the presentation

- 1 STAFF3d-spin project: multilayered nanotubes
- 2 Ru(Ox) ALD on planar substrates
- 3 ALD on high aspect ratio structures – exposure mode
- 4 HfOx nanotubes by ALD in porous membranes
- 5 Summary

# Atomic Layer Deposition (ALD) – CVD family member



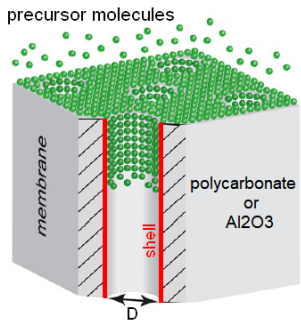
Sequential self-limiting surface chemical reactions

Sample's surface exposed **in cycles** to (for Ru):

- ① **precursor** – organometalic ( $\text{Ru}(\text{EtCp})_2$ ),
- ② Ar purge (removal of excess molecules),
- ③ **reactant**/ $2^{\text{nd}}$  precursor ( $\text{O}_2$ ),
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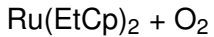
- ① **precursor** – organometallic (Ru(EtCp)<sub>2</sub>),
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- ③ **reactant**/2<sup>nd</sup> precursor (O<sub>2</sub>),
- ④ Ar purge (removal of excess molecules).

- surface chemistry-dependent (→ functionalization)
- slow, coating could be very granular
- + thickness control @ atomic scale
- + conformal coating (uneven surfaces)
- \* variety of materials: mainly oxides, but also metals (Co, Ru, Cu)

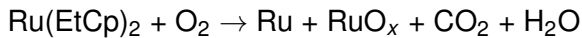
A nice ALD review: [George, Chem. Rev. 110, 111-131 \(2010\)](#)

List of materials: [Miikkulainen et al., JAP 113, 021301 \(2013\)](#)

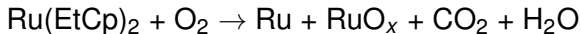
# Atomic layer deposition of ruthenium (Ru)



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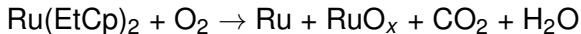


# Atomic layer deposition of ruthenium (Ru)



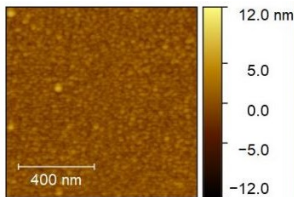
- HfOx buffer layer: [Egorov et al., \*Appl. Surf. Sci.\* \*\*419\*\*, 107-113 \(2017\)](#)
- Ru with ALD booster: [Kozodaev et al., \*J. Chem. Phys.\* \*\*151\*\*, 204701 \(2019\)](#)
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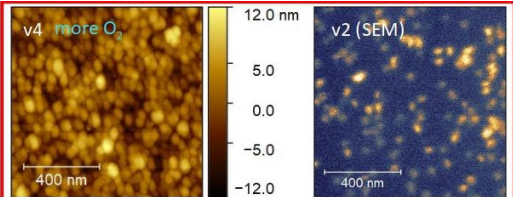


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More O<sub>2</sub>: partial oxidation of Ru, but better surface coverage (smooth, more compact layers); also temperature-dependent (here 300°C+)

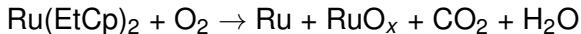


v4 on Si/HfOx (ALD)



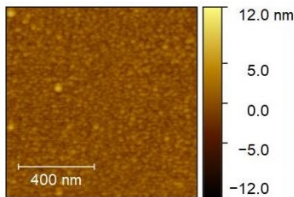
on Si/SiOx (native)

# Atomic layer deposition of ruthenium (Ru)

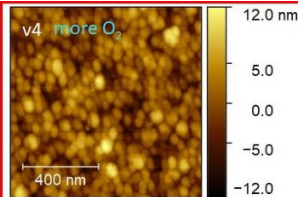


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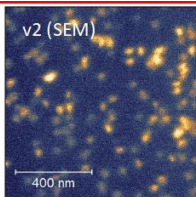
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v4 on Si/HfOx (ALD)



on Si/SiOx (native)



v2 (SEM)

H<sub>2</sub> reduction of RuO<sub>x</sub> in ALD cycle, or after the deposition (tested)

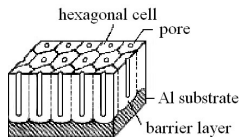
Ru/Co combination (H<sub>2</sub> used for Co deposition)

# ALD exposure mode for high aspect-ratio structures

- Surface physics: dosage = pressure · time [Langmuir = torr· $\mu$ s]

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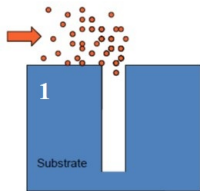
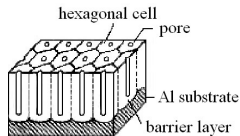
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- Pores: higher (longer) exposure to precursor gas needed



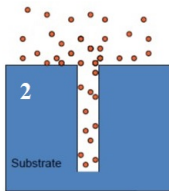


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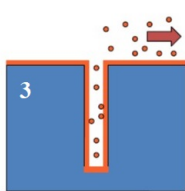
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Isolate chamber to static vacuum;  
Introduce a dose of precursor



Allow time X for precursor to diffuse into trenches, pores

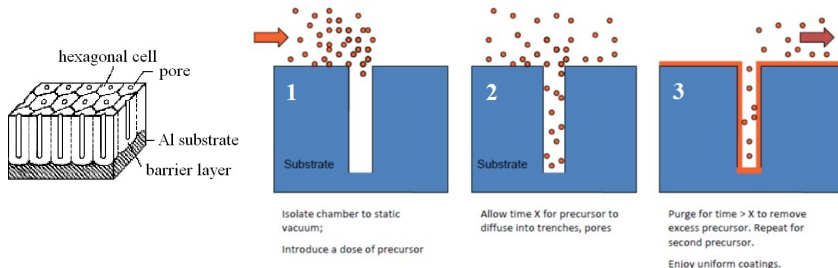


Purge for time > X to remove excess precursor. Repeat for second precursor.

Enjoy uniform coatings.

# ALD exposure mode for high aspect-ratio structures

- Surface physics: dosage = pressure · time [Langmuir = torr·μs]
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- Common solution: stopvalve (pumping stopped for a while)
- Molecules have to diffuse inside: time · temperature helps
- Thermal ALD only (not for plasma-assistance)

# Deposition of HfO<sub>x</sub> nanotubes in porous alumina

Why HfO<sub>x</sub>?

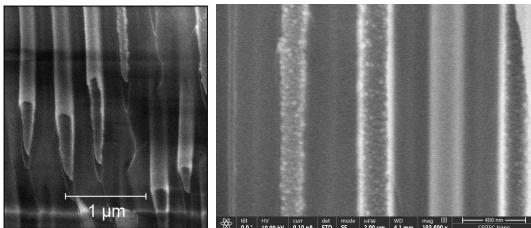
- cheaper and easier than Ru (or even Co)
- protective layer, insulating barrier (magnet/insulator/magnet)
- buffer layer for smooth Ru

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200°C, ≈ 70 s exposure for pore aspect ratio 175 ( $\phi = 200$  nm,  $L = 35$  μm)

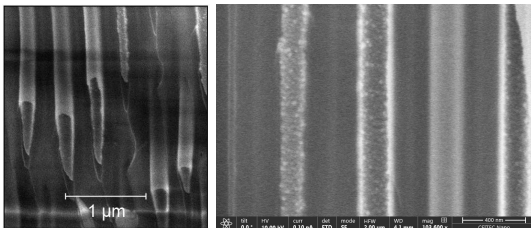


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How about other materials (Ru, Co)? Required exposure depends on:

- pore geometry (aspect ratio; dominating): **OK, it will be the same**
- temperature: **OK, it will be even higher**
- materials: **Ru, Co more difficult** (technical issues rather than chemistry)

# Summary

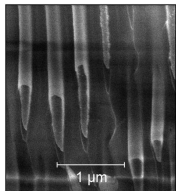
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- **Exposure mode** for nanotubes in pores – done for **HfO<sub>x</sub>**

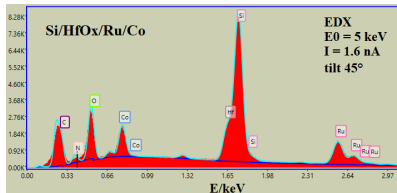
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HfO<sub>x</sub> nanotubes

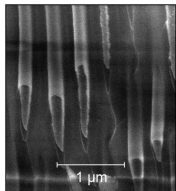


Co ALD on Ru (Si/HfO<sub>x</sub>/Ru/Co)

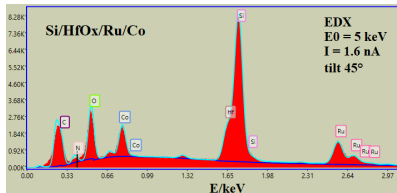
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HfO<sub>x</sub> nanotubes



Co ALD on Ru (Si/HfO<sub>x</sub>/Ru/Co)

## Plan

- Electroless plating of nanotubes (CoB or CoNiB)
- Investigation of isolated magnetic nanotubes
- Combining electroless magnetic layer + Ru spacer by ALD
- ? Co nanotubes, Co/Ru/Co by ALD



# Acknowledgements

**Thank you for your attention!**

This work was supported by the ESF under the project CZ.02.2.69/0.0/0.0/19\_074/0016239.



EUROPEAN UNION  
European Structural and Investment Funds  
Operational Programme Research,  
Development and Education



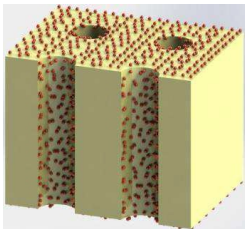
We acknowledge CzechNanoLab Research Infrastructure supported by MEYS CR (LM2018110).

STAFF3d-spin project: <http://magnetism.ceitec.cz/staff3d-spin/>  
(slides of presentations, updates, ...)

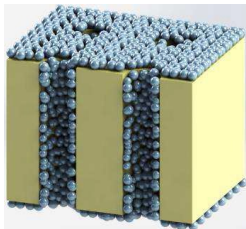
# Electroless deposition of nanotubes

Conformal coating of modified walls of porous template

- Silanization of pore walls ( $\text{Al}_2\text{O}_3$  only) – APTES
- Sensitization of pore walls –  $\text{SnCl}_2$
- Activation –  $\text{PdCl}_2$  (Pd seeds)
- Deposition – immersion in a plating bath
- Removal top/bottom layer, template dissolution



Template with seed particles



Initial growth

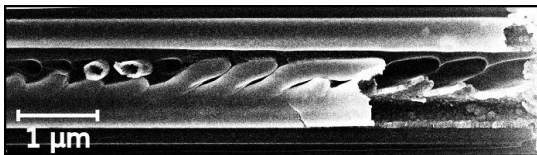
Plating bath:

- metal salt ( $\text{CoSO}_4$ )
- reducing agent (DMAB)
- stabilizing agent (sodium citrate)

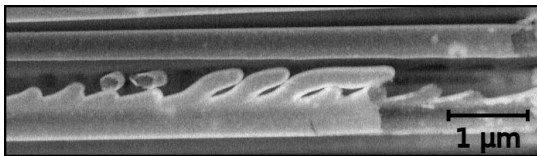
Images from *ECS Trans.* **64** (31), 39-48 (2015).

# CoNiB electroless plating in $\text{Al}_2\text{O}_3$ porous template

Commercial template with pore branching close to the end;  
cross-section, template and deposit (partially) broken



Secondary electrons



Back-scattered electrons (less charging, more material contrast)

Electroless plating '*replicates*' the pore shape, including defects.

## Towards nanotubes: deposition routes

How about nanotubes when the porous template is from  $\text{Al}_2\text{O}_3$ ?!

- metallize the pores (Ru ALD or electroless plating below)

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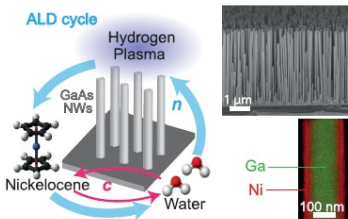
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[ $\text{Fe}(\text{Cp})_2$  = ferrocene precursor available], similar to Ni from  $\text{Ni}(\text{Cp})_2$   
*Giordano et al., ACS Appl. Mater. Interfaces 12, 40443-40452 (2020)*

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- physical vapour deposition on vertical nanowires – Dr. Kolíbal
- electroless plating, also seed metallic layer for ALD in nanopores (acquired chemicals for CoB, CoNiB)